

Biophysical Characterization of Aptamer–small molecule Interactions

MicroScale
Thermophoresis

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2bind
molecular interactions



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Technology

MicroScale Thermophoresis (MST) is an optical fluorescent method. It records the changes in fluorescence of a target molecule as a function of temperature and the concentration of the cognate ligand molecule. Thus, in MST, the target must be fluorescent and the ligand must not be fluorescent in the same wavelength range as the target. The fluorescence changes are the result of two effects:

First, molecules exhibit directed movement along temperature gradients in solution, an effect called thermophoresis (Duhr & Braun, 2006). Such thermophoretic movement of a molecule is characterized by its size (hydrodynamic radius), its surface charge, and its hydration shell. All three parameters can be affected by binding of a ligand molecule. Thus, a concentration-dependent titration of the target molecule's ligand induces changes in the thermophoretic movement. These changes translate in quantifiable, spatial fluorescence changes, which are easily tracked optically via the target fluorescence.

Second, fluorescence is a function of the temperature. Thus, the fluorescence of the target molecule varies with the temperature (Baaske *et al.*, 2010). This temperature-dependence is additionally affected by the local molecular surroundings of the fluorophore. Thus, binding of a ligand molecule to the fluorescent target can change the chemical environment of the target fluorophore and thus the overall detected temperature-dependence of the fluorescence. This effect is known as TRIC (temperature-related intensity change; Gupta, Duhr & Baaske, 2018).

The thermophoresis and TRIC signals are additive and thus both contribute to the high sensitivity and robustness of MST measurements towards molecular binding events of all kinds. Thus, MST can be used for determining the affinity and binding strength of almost any kind of molecular interaction with very low sample consumption and very high sensitivity.

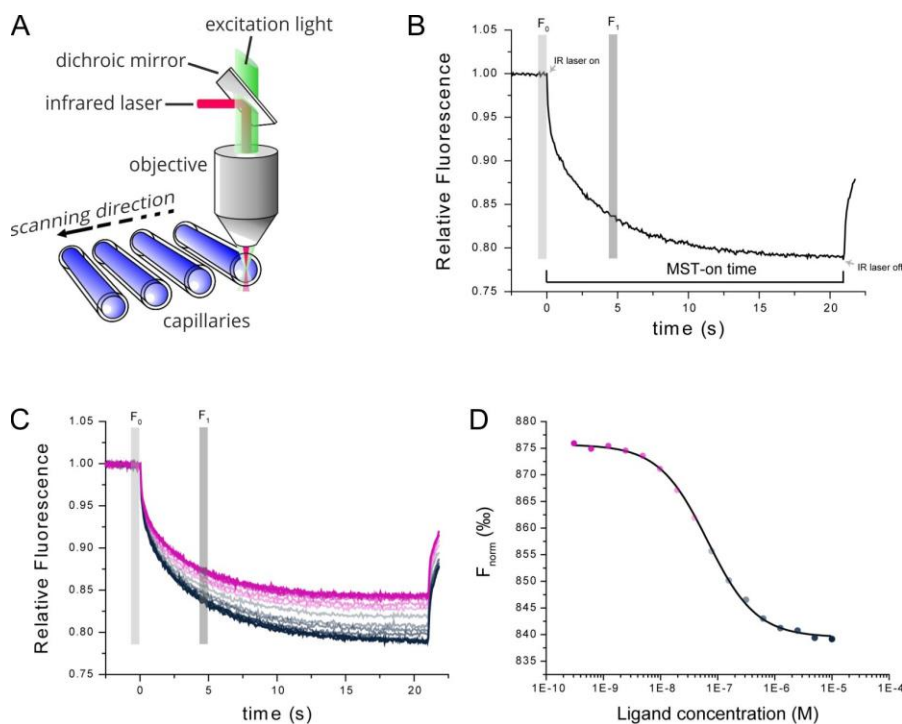


Figure 1. Technical MST setup. (A) MST measurements take place in small glass capillaries. Infrared and fluorescence lasers are used for generation of the MST effect and sample tracking. (B) TRIC and thermophoresis together account for a time-dependent change in fluorescence upon infrared-heating of the sample capillaries. (C) Multiple MST traces are recorded for different mixture ratios of target and ligand molecules. (D) Dose-response analysis of the MST traces allows for determination of the steady-state affinity of the target-ligand interaction.

Features

- ✓ Wide molecule size range
- ✓ Real-time quality controls
- ✓ No immobilization required
- ✓ Very short analysis time
- ✓ Free choice of assay buffer
- ✓ Low sample consumption
- ✓ Wide affinity range
- ✓ Wide temperature range

Advantages

- Analyze molecules from 100 Da up to 1 MDa
- Online aggregation, precipitation, and sticking controls
- Measurement is done truly in solution
- High analysis speed enables high throughput
- Biological liquids such as serum or cell lysate possible
- Minimum of only 6 μ l per sample, nM concentrations
- Affinities can be analyzed in the pM-mM range
- Analysis possible from 20°C to 45°C

Introduction

Aptamers are single-stranded nucleic acid molecules with a well-defined, folded tertiary structure. With that structure, aptamers have the potential to recognize various classes of target molecules independent of their size. Even challenging targets such as small molecules can be bound by aptamers with high affinity and specificity (Kim & Gu.2014). In general, an analysis of the interaction between aptamers and small molecules in detail is challenging. It is often complicated to access their basic binding parameters such as their affinity using many biophysical methods (Entzian & Schubert. 2016). Here, we demonstrate how MST-assay can be used to map the binding site of the ATP aptamer DH25.42 and to set up an assay format that allows a fast and precise affinity characterization of aptamer-small molecule interactions. For mapping the specific binding site, the binding affinity of the DH25.42 aptamer towards multiple ATP-analogues (ADP, AMP, SAM, dATP, CTP, GTP and adenine) were measured and compared.

Assay type	MST
Target	DH25.42 aptamer
Labeling	Cy5
Ligand	ATP, SAM, AMP, ADP, Adenine, GTP, CTP, dATP

Results

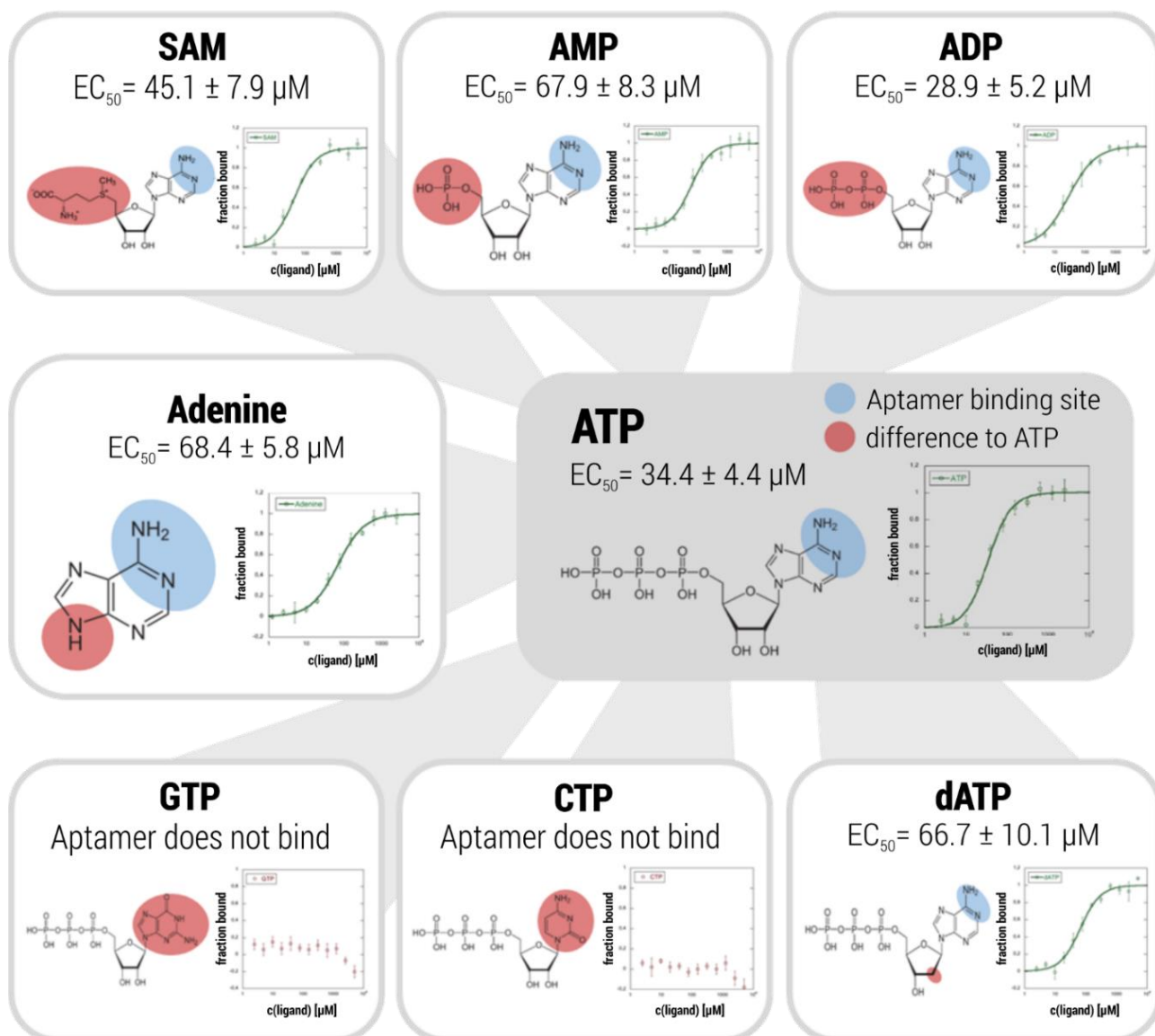


Figure 2. Binding-site mapping of an ATP-aptamer. ATP, ADP, AMP, SAM, dATP, CTP, GTP and adenine were titrated against the ATP aptamer DH25.42. Dose-response data were fitted to the hill equation (EC_{50}). Error bars show the standard deviation derived from three repeats. The blue shaded area indicates the binding site of the aptamer on the adenine group of ATP, the red shaded area shows the molecular difference to ATP.

The ATP-analogues ADP, AMP and SAM differ in the 5'-position of the ribose compared to ATP. Collating the detected affinities of ADP ($28.9 \pm 5.3 \mu\text{M}$), AMP ($67.9 \pm 8.3 \mu\text{M}$) and SAM ($45.2 \pm 7.9 \mu\text{M}$), we suggest that this position has no or only minor influence on the binding behavior of the aptamer (Figure 2). dATP, lacking the hydroxy group at C2 of the ribose, was also bound by the aptamer with a slightly reduced affinity ($66.7 \pm 10.1 \mu\text{M}$). In contrast, CTP, with a purine instead of the ATP-pyrimidine ring, was not detected by the aptamer, implying that adenosine represents the main aptamer binding site. This indication is further strengthened as adenine is bound by the aptamer with an affinity of $68.4 \pm 5.8 \mu\text{M}$. It is worth mentioning that GTP, which has a high overall similarity to ATP remains unbound by the aptamer narrowing down the binding site.

Conclusion

Considering the binding affinity of DH25.42 aptamer to various ATP-analogues, we successfully established a MST set up that allows a fast and precise kinetic characterization of any aptamer-small molecule interaction. Additionally, we could map the binding site of the ATP aptamer DH25.41 down to a specific molecular moiety at the adenine of the ATP molecule.

Literature

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